A New Gun Barrel Erosion Reducer

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ROSION of gun tubes by hot propellant gases has long been a problem. A significant reduction in erosion was achieved when titanium dioxide in a wax carrier was coated on a cloth and lined inside the top section of the cartridge case around the periphery of the propellant charge. This wear reducing additive method is basically a Swedish development and was patented in this country by Jacobson and Ek.1

We have been pursuing this problem from two aspects: the mechanism responsible for this reduction, and evaluation of materials other than TiO₂.

A silica-wax system was as good an erosion reducer as the TiO₂-wax combination. Its effectiveness was attributed to the large surface area of the silica particles and its high heat content. This finding led to the evaluation of commercially available metallic silicates. A magnesium silicate, talc, was significantly more effective than TiO₂. This note presents the data obtained to date with the various silicates, particularly talc.

Laboratory Studies

A standard 200-cm³ closed bomb was modified by placing an erosion sleeve of gun metal in a barrel segment attached to the bomb's gage housing (Fig. 1).2 Propellants with or without additives are burned in the bomb. A rubber stopper in lieu of a projectile is accelerated through the erosion sleeve and the barrel as the propellant burns. The propellants, with a single-perforated geometry, are fired at approximately 16,000 psi. The weight loss of the erosion sleeve for each series of firings is a measure of the extent of erosion. All pressures are measured with a strain gage and recorded with an oscillograph. A number of silicates were evaluated including kaolin, kaolinite, muscovite mica, feldspar, and talc. The latter is a magnesium silicate, whereas the others are aluminum silicates. The aluminum silicates were in no case superior to the talc.

An experiment was designed to study the effectiveness of tale vs TiO₂. Tales that differed in particle size were obtained for this study. The TiO2 had already been standardized as a wear-reducing agent. These additives were blended with four commercially available waxes ranging in melting point from 120° to 180°F. The additive/wax ratios, in turn, were $\frac{45}{55}$, $\frac{22}{78}$ and $\frac{11}{89}$. Five successive firings were made with

Table 1 Erosion study of talcs vs TiO2 in wax carriera

					Tales					
	TiO ₂ , 0.6 μ		Supreme, 2.3 μ			Mistron vapor, 0.6 μ				
0	45	99	11	45	99	11	45	22	11	
-	40	22	11	40	22		10			
10.2	5.3	8.6	12.1	1.5	6.7	9.0	3.4	4.9	6.0	
5.7	4.5	7.6	9.4	5.3	7.2	9.0	4.0	4.8	6.7	
9.3	4.8	7.5	10.3	5.1	6.5	8.7	3.5	4.7	5.7	
8.4	5.0	7.6	11.0	3.3	5.3	6.9	1,5	3,2	4.0	
	$\frac{5.7}{9.3}$	0 45 rrier 10.2 5.3 5.7 4.5 9.3 4.8	0 45 22 rrier 10.2 5.3 8.6 5.7 4.5 7.6 9.3 4.8 7.5	0 45 22 11 rrier 10.2 5.3 8.6 12.1 5.7 4.5 7.6 9.4 9.3 4.8 7.5 10.3	0 45 22 11 45 rrier 10.2 5.3 8.6 12.1 1.5 5.7 4.5 7.6 9.4 5.3 9.3 4.8 7.5 10.3 5.1	0 45 22 11 45 22 rrier 10.2 5.3 8.6 12.1 1.5 6.7 5.7 4.5 7.6 9.4 5.3 7.2 9.3 4.8 7.5 10.3 5.1 6.5	TiO ₂ , 0.6 μ Supreme, 2.3 μ 0 45 22 11 45 22 11 rrier 10.2 5.3 8.6 12.1 1.5 6.7 9.0 5.7 4.5 7.6 9.4 5.3 7.2 9.0 9.3 4.8 7.5 10.3 5.1 6.5 8.7	TiO ₂ , 0.6 μ Supreme, 2.3 μ Mist of Supreme, 2.3 μ Supreme, 2.3 μ 1 45 rrier 10.2 5.3 8.6 12.1 1.5 6.7 9.0 3.4 5.7 4.5 7.6 9.4 5.3 7.2 9.0 4.0 9.3 4.8 7.5 10.3 5.1 6.5 8.7 3.5	TiO ₂ , 0.6 μ Supreme, 2.3 μ Mistron v. 0 45 22 11 45 22 11 45 22 17 45 22 rrier 10.2 5.3 8.6 12.1 1.5 6.7 9.0 3.4 4.9 5.7 4.5 7.6 9.4 5.3 7.2 9.0 4.0 4.8 9.3 4.8 7.5 10.3 5.1 6.5 8.7 3.5 4.7	

a 30 g of propellant M2 ($T_v = 3320$ °K) fired with 4 gm of blend of additive/wax carrier consisting of 45% additive/55% wax, 22% additive/78% wax, or 11% additive/89% wax. The erosion is expressed as milligrams wear for five successive shots.

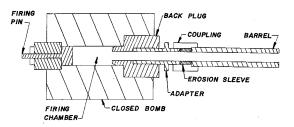


Fig. 1 Sketch of erosion apparatus.

each sample utilizing M2 Propellant (isochoric flame temperature, 3320°K) along with the waxes as controls. The results of this experiment are presented in Table 1. There were significant differences among the waxes as well as between the talcs and the TiO_2 . Moreover, the 0.6- μ average particle size talc was significantly superior to the 2.6- μ talc. The more additive, and the less wax, used in the sample likewise resulted in a significant reduction in the wear of the erosion sleeve.

The insertion of a wear-reducing liner in the cartridge case for small arms ammunition would be impracticable because of the high-loading density of these rounds and the smallness of the cartridge cases. Consequently, the wear-reducing additive, talc, was incorporated in the propellant itself. The data presented in Table 2 shows that the incorporation of 2% talc in the propellant base grain was an effective wearreducing agent for both 7.62 and 20-mm small arms ammuni-

Weapon Firings

In order to corroborate the favorable laboratory data for talc, this complex silicate has been evaluated in three systems to date. In the 105-mm gun, M68, the talc/wax mixture $(2.3\% \pm 0.5\%)$ of the propellant charge weight) was coated on a rayon cloth and lined inside the top portion of the cartridge case around the periphery of the propellant charge. No wear occurred in the gun tube from firing 50 rounds, each, of the HEAT, M456 and APDS, M392 projectiles with M30 Propellant. Fifty, each, of these same two rounds with the standardized TiO₂/wax liner gave wear values of 3 and 4 mils, respectively.3

One hundred rounds were fired in the 175-mm gun M113, with the XM1 additive jacket slipped over the zone 3 bag of the M86E1 propelling charge. (This jacket weighed $2.9\% \pm$ 0.2% of the M6 Propellant charge weight.) The rate of wear was 40 mils/100 rounds using an additive jacket containing a talc/wax mixture.4 Previous data showed that the average rate of wear was 49 mils/100 rounds when the additive jacket contained a titanium dioxide/wax blend.

Barrel erosion is a serious problem in the 40-mm M75 launcher. The life of the launcher was approximately 1500 rounds before the hot M2 propellant gases eroded away the

Table 2 Erosion from small arms propellants with and without tale added

	prop	-mm ellant 138M ^a	20-mm propellant CR7814 ^b		
	Control	2.5% talc (0.6 µ)	Control	2.1% tale (0.6 \(\mu\))	
Web, in. SP	0.0139	0.0133	0.0208	0.0227	
Measured heat of explosion, cal/g Relative force at	896	898	879	808	
max pressure At 70°F	100.0	97.5	100.0	94.4	
At -65°F	100.0	97.9	100.0	94.2	
Erosion, mgc	4.8	0.5	5.0	1.2	

a The calculated thermochemical properties of IMR8138M are: isochoric

b Melting point at 1 atm.

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flame temperature, 2810°K; force, 324,800 ft-lb/lb.

b The calculated thermochemical properties of CR7814 are: isochoric flame temperature, 2590°K; force 308,800 ft-lb/lb. 30 g of propellant fired to give maximum pressure of 16,000 psi.

forcing cone and the commencement of rifling. However, one launcher tube had over 5300 40-mm M385 rounds fired through it and still had not reached its condemnation limit when a polyethylene bag containing talc was placed on top of the propelling charge in the M169 case.⁵

Small Arms Program

An extensive program is now planned for the evaluation of 0.6- and $2.3-\mu$ average particle size talcs in small arms ammunition. Four CR7814 propellant lots containing 1 and 2%, respectively, of both particle size talcs are being procurred from the duPont Co. These four special propellant lots will be loaded at Lake City Army Ammunition Plant into cartridges, 20-mm T.P. M55A2, as well as a control lot of standard WC870 ball propellant. The ammunition plant will fire temperature coefficient and erosion tests with both the experimental and the control ammunition.

References

¹ Jacobson, D. E. and Ek, S. Y., "Wear Reduction Additives," U.S. Patent 3,148,620, Sept. 15, 1964.

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Nov. 1965, Picatinny Arsenal, Dover, N.J.

³ Fletcher, D. H., "Final Report on Engineering Design Test of Ammunition Additive for 105mm Gun, M68," DPS-2368, May 1968, Aberdeen Proving Ground, Aberdeen, Md.

⁴ Norton, J. J., "Product Improvement Tests of Charge, Propellant, 175mm M86A1 (Talc Additive Jacket)," D&PS Firing Record P-74726, Sept. 1967, Aberdeen Proving Ground, Aberdeen, Md.

⁵ Decker, D. B., "A Test to Determine the Cause of Barrel Erosion in the 40mm M75 Launcher," TR3576, June 1967, Picatinny Arsenal, Dover, N.J.

Advanced Pressurization System for Liquid Rockets

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Nomenclature

 $\begin{array}{lll} c^* & = & \text{characteristic exhaust velocity} \\ g & = & \text{acceleration of gravity, } 32.174 \text{ ft/sec}^2 \\ I_s & = & \text{specific impulse, lbf-sec/lbm} \\ MR & = & \text{mixture ratio} = W_o/W_f \\ P & = & \text{pressure} \\ R & = & \text{gas constant, ft-lb/lb}^o \\ t_b & = & \text{burning time, sec} \\ T & = & \text{temperature, } ^\circ R \\ V_v & = & \text{velocity and volume, respectively} \\ W & = & \text{weight} \\ \end{array}$

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Subscripts

BO,O = burnout and gross, respectively

c = chamber (injection end)

f,g,o = fuel, gas, and oxidizer, respectively

NPSH =net positive suction head

p,PL = propellant and payload, respectively

Introduction

THIS Note describes advanced systems studies concerned with using the main propulsion system of a liquid rocket to provide some of the so-called auxiliary functions, thereby reducing pressurization system inert weights and resulting in system simplification. The engine would generate vaporized propellants for tank pressurizing. The engine can also burn most of the tank pressurizing gases and the liquid residuals presently unavailable. The net result is an increase in payload of 2250 lb (2%) for the Apollo mission (Table 1).

In the existing Saturn S-IVB stage, the tanks are initially pressurized by helium from a ground supply. During engine operation, the fuel tank is pressurized by gaseous hydrogen (GH₂) bled from the thrust chamber and the oxidizer tank is pressured by stored He, which is heated by an engine heat exchanger. For the Apollo mission, the first burn places the S-IVB in a 100-naut-mile earth orbit. During earth orbit coast, the fuel tank is vented continuously to 20 psia. The

Table 1 Potential payload gains for Apollo mission, S-IVB with J-2S

	Current d	esign	Advanced system		
	W, lb	MR	W, lb	MR	
Useful propellants	(225,300)	5.0	(226, 591)	4.915	
Main stage burn	225,300	5.0^a	220,968	4.943^{b}	
J-2S tank repressuriza-					
tion prior to 2nd					
burn			4,277	4.0c	
J-2S, end $2nd$ burn			1,346	4.0°	
Boiloff in orbit	3,880	0	3,880	0	
Tank pressurization gas					
Gaseous propellant	783	0.71	1,786	2.5	
Helium (at burnout)	218		28		
Liquid residuals	1,269	0.46	1,269	0.46	
Propellant utilization	,		·		
reserve	136	0	136		
Engine and ducts	1,133	0.55	1,133	0.55	
Total unusable gases	1,001	0.94	$^{'}454^{d}$	2.5	
Total unusable liquids	1,269	0.46	206^{d}	0	
Performance reserve	, -				
liquids ^e	3,200	5.0	1,969	4.0	
Performance reserve	-,	*	-,		
gases ^e	0		1,360	2.5	
Helium bottles	600		0		
O ₂ /H ₂ burner	50		0		
Total inert (nominal	00			• • • •	
flight)	6,120		3,989		
Total effect on payload	0,220		3,000		
Inert weight decrease			+2,131		
Increase in usable W_p			+ 97		
Decrease in engine MR			+ 161		
I_s during repressur-			1 101		
ization			- 61		
I_s (nominal flight)			- 40		
Gravity loss			- 30		
Net payload gain, ΔW_{PL}			+2,258		
Dry weight, including IU	29,257		28,607		
Stage inert weight	34,727		32,596		
W_0 , at ignition	377,522		378,940		
W_{BO} , burnout weight	148,342		148,469		
W_{PL} , payload	113,615		115,873		
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 $^{{}^{}a}I_{s} = 434.0$, F = 240,000 lb.

e Total performance reserve = propellant for 0.75% of vehicle ideal $\Delta V = 302$ fps.

 $b I_8 = 434.34$, F = 240,000 lb. $c I_8 = 421.0$, F = 50,000 lb.

 $^{^{\}circ}$ $I_3 = 421.0$, F = 50,000 lb. d = 25% of gases + 70 lb LH_2 trapped in tank + propellant utilization reserve LH_2 .